

Doctoral program Industrial engineering, cycle XL

Advanced adsorbent materials for low-grade thermal cooling applications

PhD student: Sondes GUESMI

**Supervisor : Prof. Luigi
CALABRESE**

Context & Motivation

Motivation

- Cooling & refrigeration account for **>10%** of global electricity
- Rising demand with climate change and digitalization
- Generation of **greenhouse gas emissions**
- Large availability of **low-grade thermal energy** ($\approx 40\text{--}80\text{ }^{\circ}\text{C}$)

International Energy Agency: IEA

Research gap

- Adsorption cooling uses heat
- Performance limited by adsorbent **material**
- **Need:** High water uptake+ low regen temperature



The system works through adsorption and desorption cycles, where **refrigerants** (water, methanol, ammonia..) are captured and released by **adsorbent materials** (porous materials) using heat.

Adsorption Phase: The adsorbent captures refrigerant vapor, releasing heat.

Desorption Phase: External heat drives the refrigerant to be released from the adsorbent material.

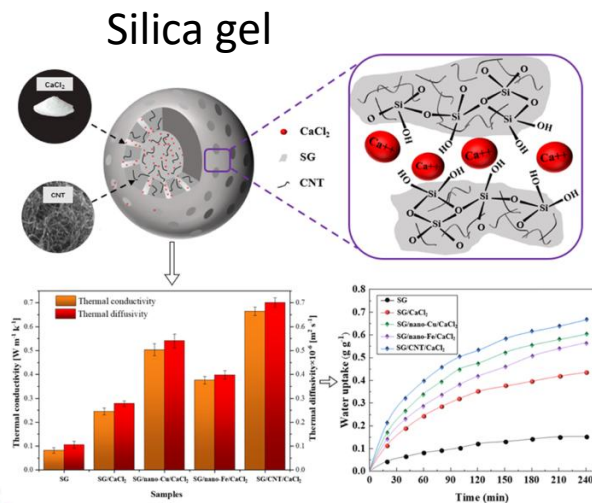
Cooling Cycle: The adsorbent captures refrigerant vapor, releasing heat.

Regeneration: The adsorbent captures refrigerant vapor, releasing heat.

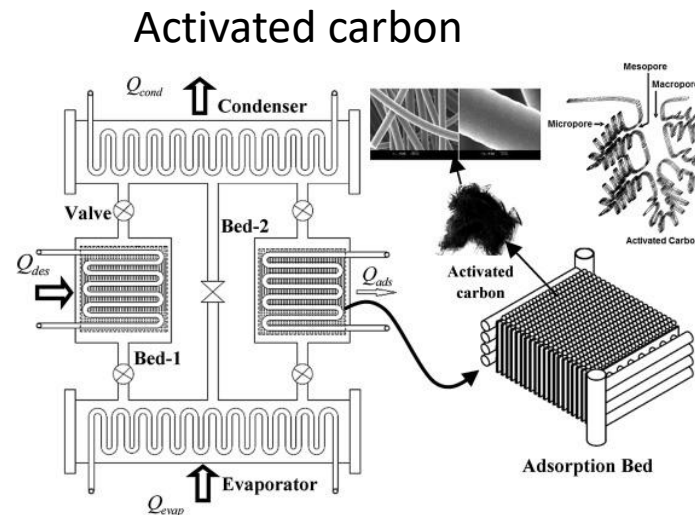
Key properties include:

- ❖ High surface area, high porosity
- ❖ Thermal & mechanical stability
- ❖ High adsorption capacity to store a significant amount of refrigerant
- ❖ Fast adsorption/desorption to enable efficient cycling

Examples:

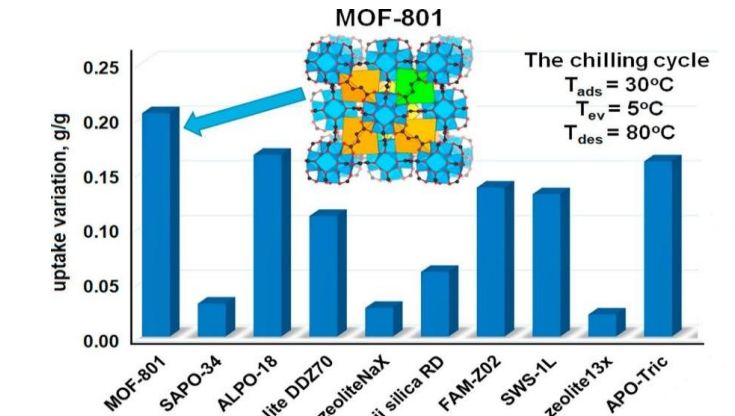


Wenhao Xie et al. Journal of Sol-Gel Science and Technology (2023)

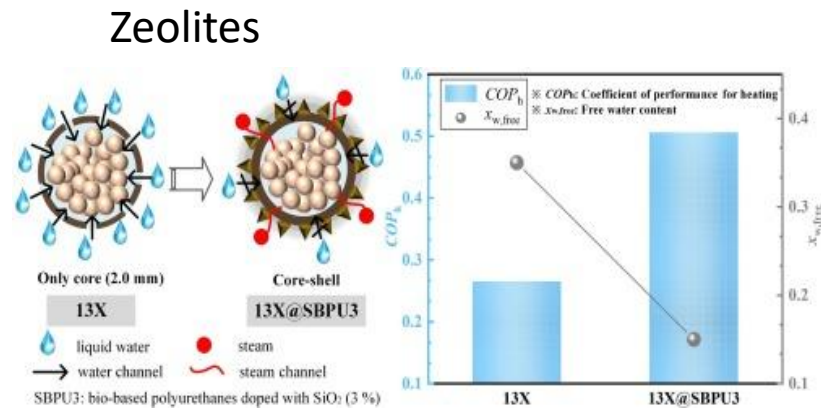


W. Fan et al. Energy 102 (2016)

Metal-organic frameworks (MOFs)



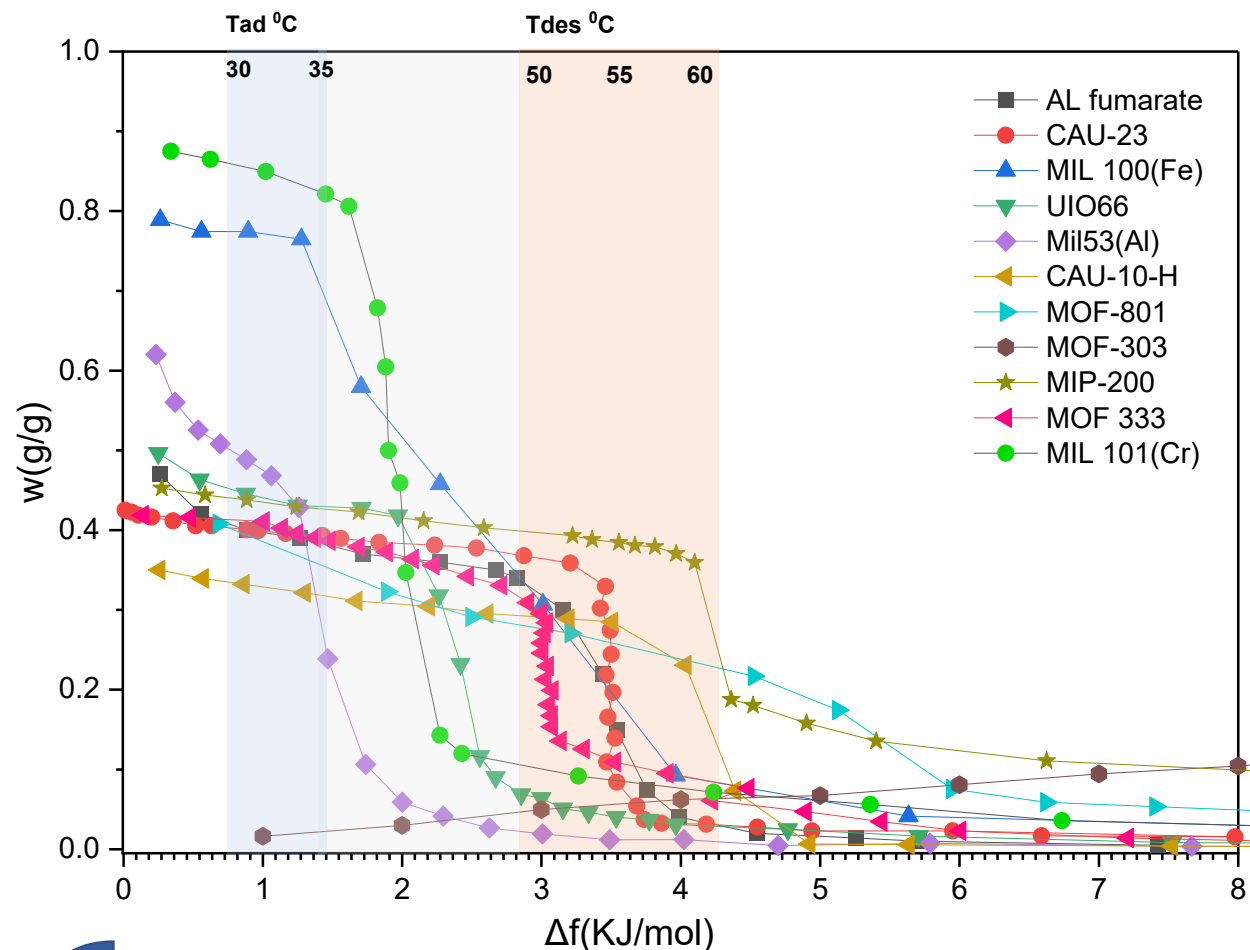
M.V. Solovyeva et al. Energy Conversion and Management (2018)



X. He et al. energy Conversion and Management (2024)

MOFs selection

Which MOFs can efficiently operate under low-grade temperature conditions?



$\Delta F1$	$\Delta F2$	Ranking	MOFs	ΔW - adsorption
1.1	3.2	1	MIL-101(Cr)	0.7574
1.1	3.2	2	MIL 100(Fe)	0.5722
1.1	3.2	4	Mil53(Al)	0.4486
1.1	3.2	5	UIO-66(Zr)	0.3952



MIL-101(Cr), MIL-100(Fe), MIL-53(Al) and UIO-66(Zr) are the most suitable for adsorption cooling at low grade temperature.

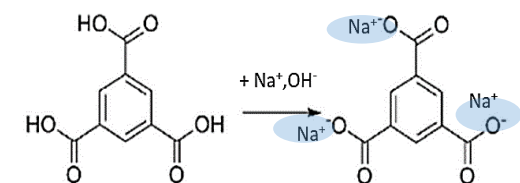
Green synthesis strategy

Green synthesis of MIL-100(Fe)

Solution A

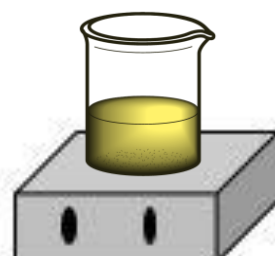


3NaOH in H_2O + H_3BTC



Trimeric acid(H_3BTC)

Solution B

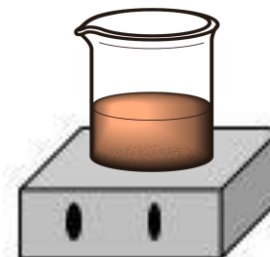
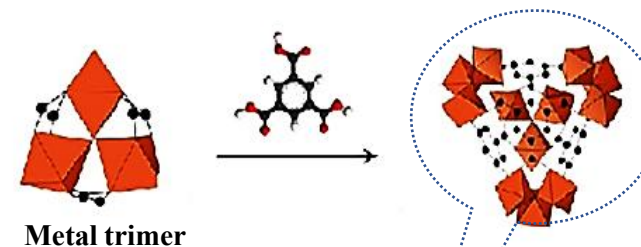


$\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ in H_2O

Add B in A



- Green synthesis (solvent water)
- Room temperature synthesis
- Simple and easy to operate



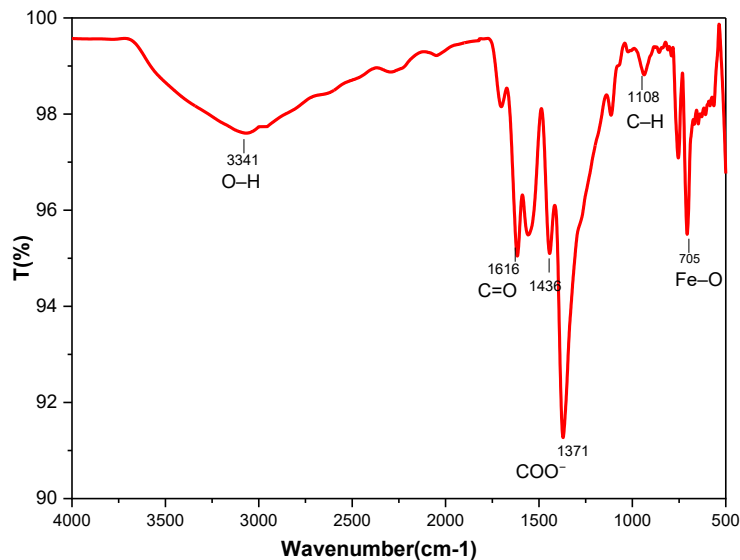
Filtration
washing, drying



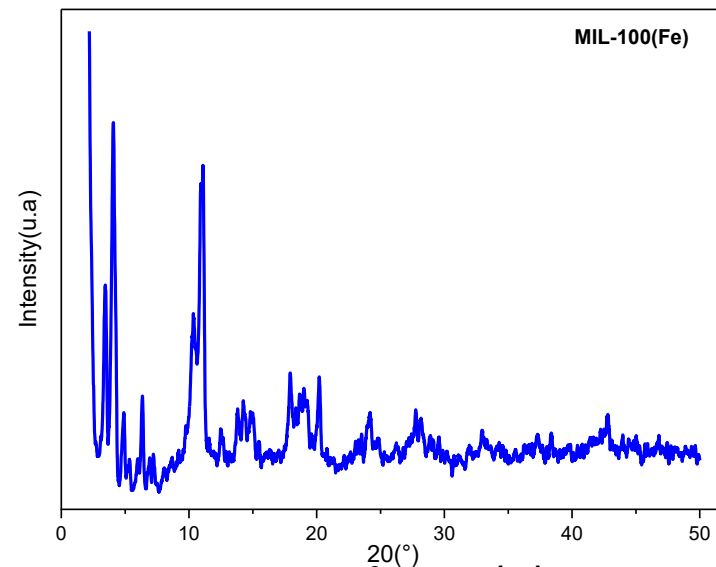
Formation of MIL-100(Fe)

MIL-100(Fe)

Characterization of MIL-100(Fe)



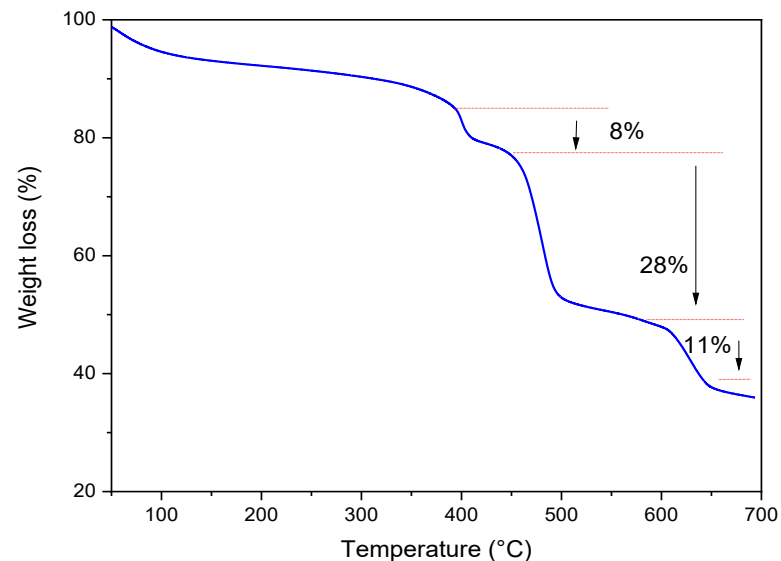
FTIR spectrum of MIL-100(Fe)



XRD spectrum of MIL-100(Fe)

Textural parameters of MIL-100(Fe) from DFT analysis (N₂, 77 K)

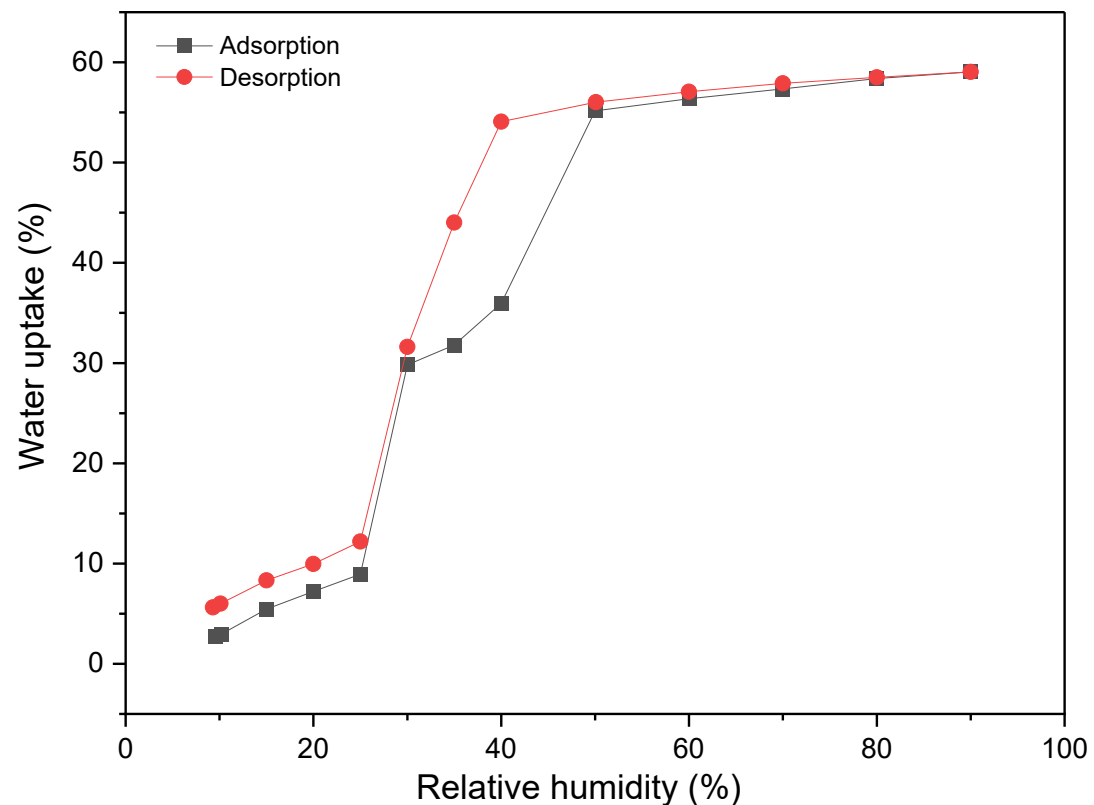
Sample	Surface area (m ² g ⁻¹)	Total pore volume (cm ³ g ⁻¹)	Mesopore volume (cm ³ g ⁻¹)
MIL-100(Fe)	1549.7	0.579	0.18



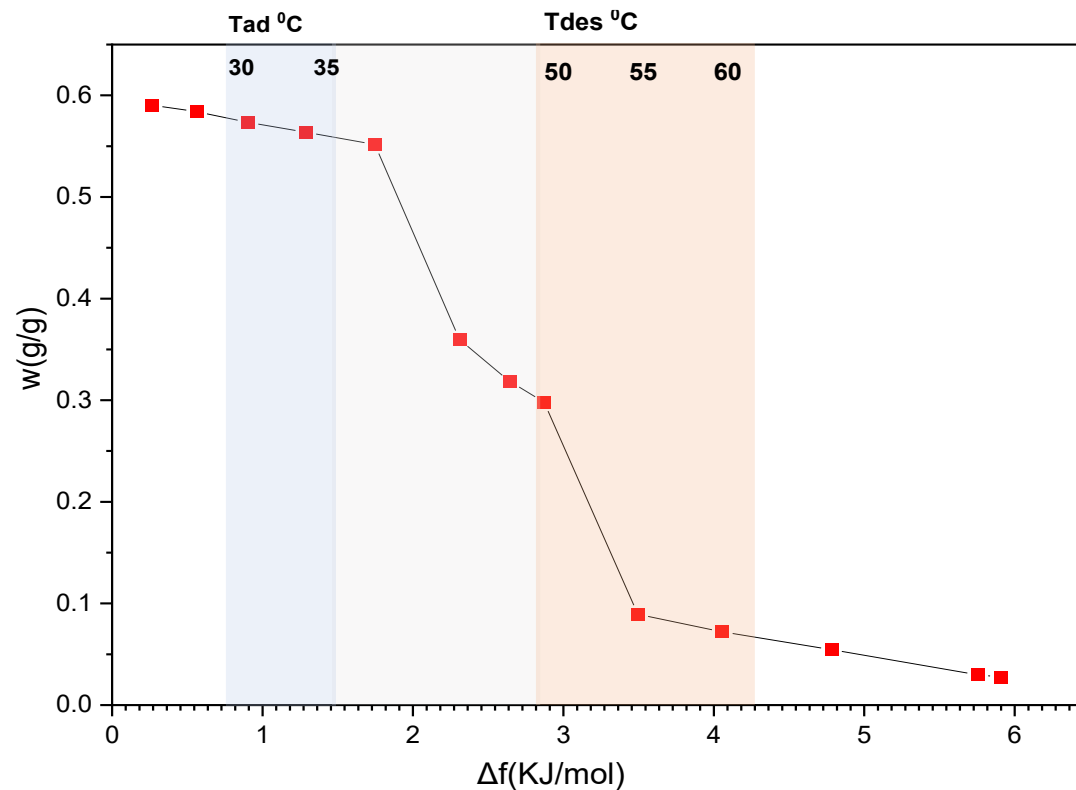
TGA/DTG curves of MIL-100(Fe)

- ✓ FTIR and XRD confirm successful formation and crystallinity of MIL-100(Fe).
- ✓ BET reveal high surface area.
- ✓ TGA shows good thermal stability up to ~300 °C.

Water Sorption Performance



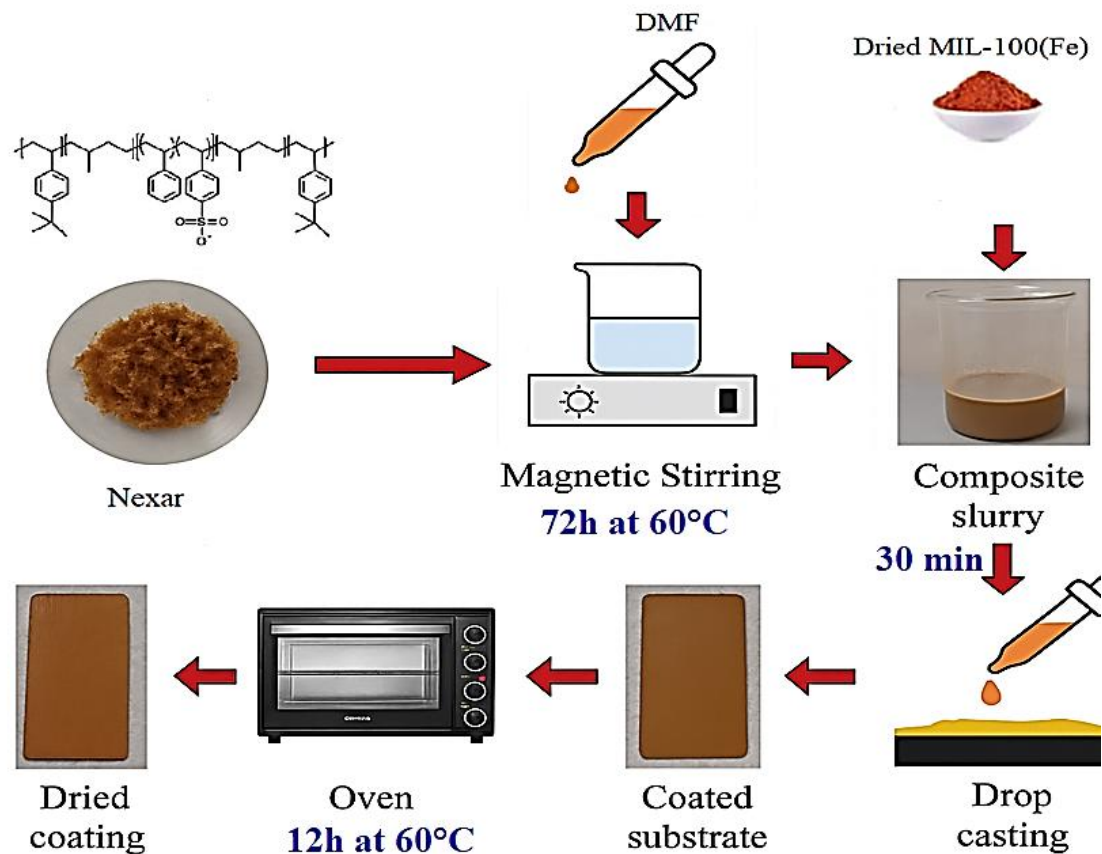
Water Vapor Isotherm of MIL-100(Fe) at 30°C



Thermodynamic operating window of MIL-100(Fe)

- ✓ Fast uptake at moderate RH confirms strong hydrophilicity.
- ✓ Thermodynamically suitable for low-grade adsorption cooling

MIL-100(Fe)-Polymer Composite Coatings



Composite coating characterization

❖ **Scratch Test:** Surface resistance

❖ **Impact Test:** Toughness

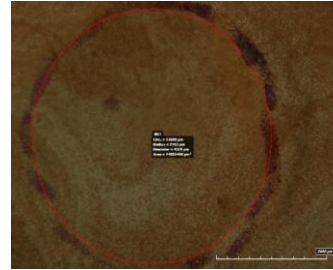
❖ **Pull-Off Test:** Adhesion strength

Coating performance

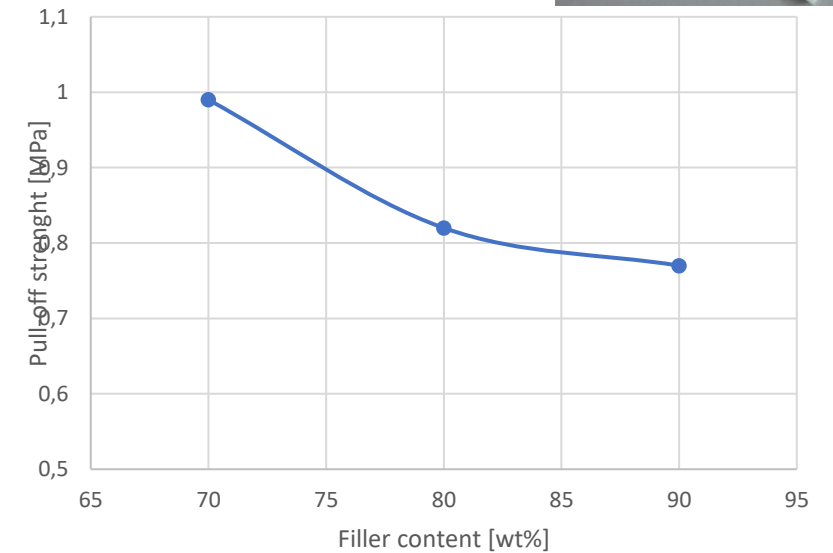
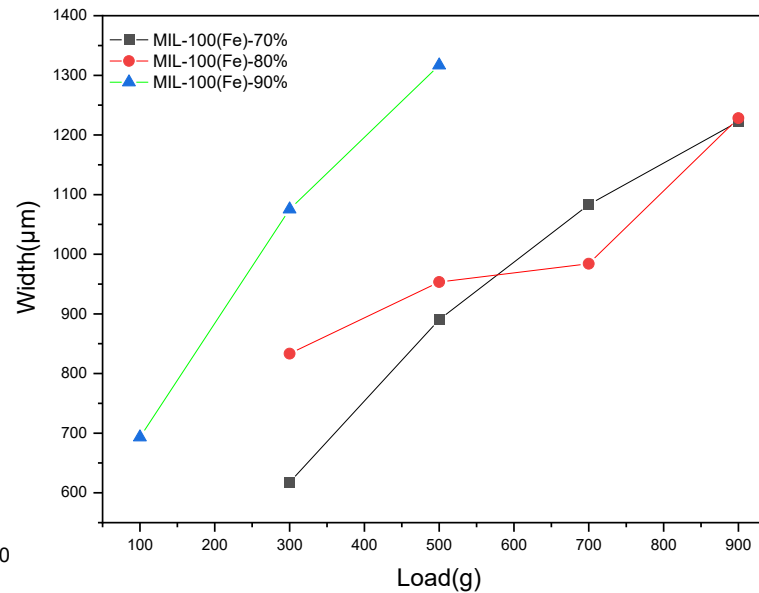
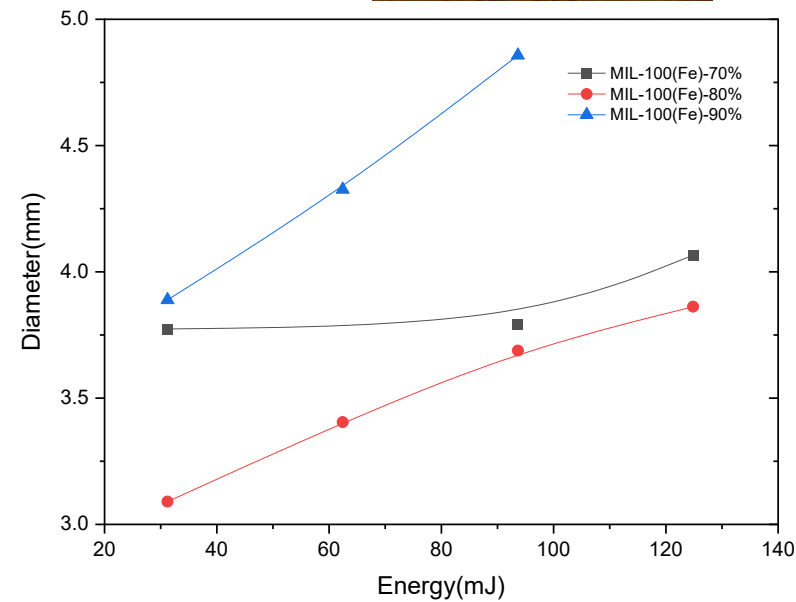
Scratch test



Impact test



Pull-Off Test



Mechanical tests show acceptable coating performance across different MOF loadings and further optimization is in progress.

Conclusions and Outlook

- ✓ Green and simple synthesis successfully achieved
- ✓ Promising water sorption demonstrated
- ✓ Composite coatings show acceptable mechanical performance
- Evaluation of the **sorption performance of composite coatings**
- Cycling stability assessment
- Heat exchanger integration

Thank you for your attention



Acknowledgments



Università
degli Studi di
Messina

DIPARTIMENTO DI INGEGNERIA



Consiglio Nazionale
delle Ricerche



 Sorption
Technologies